Overview of MCell Methods

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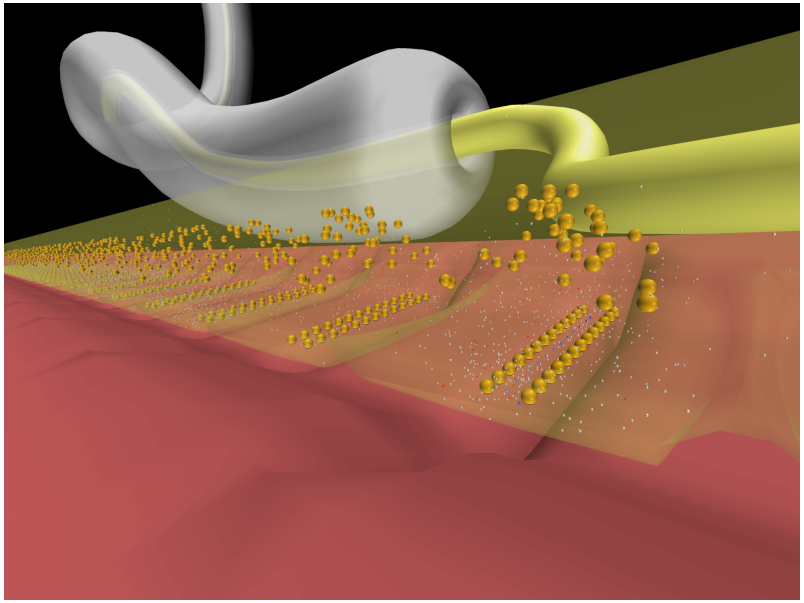
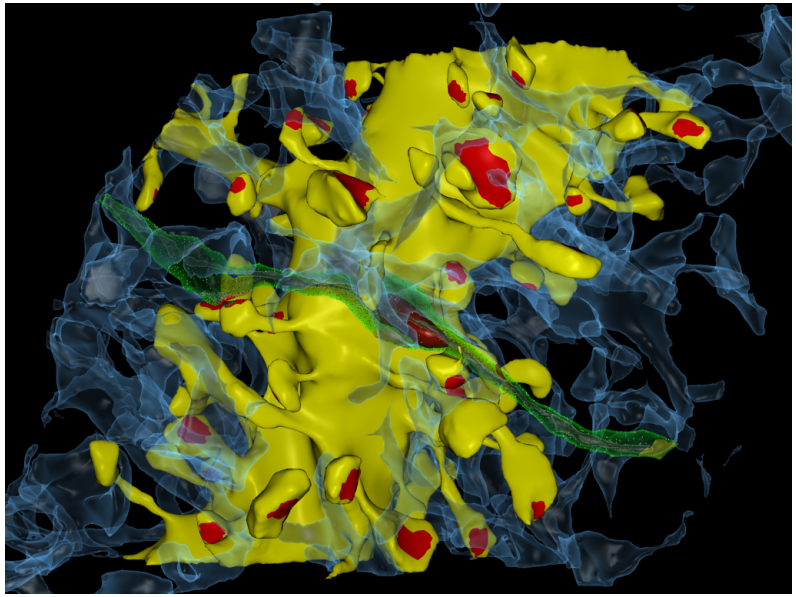
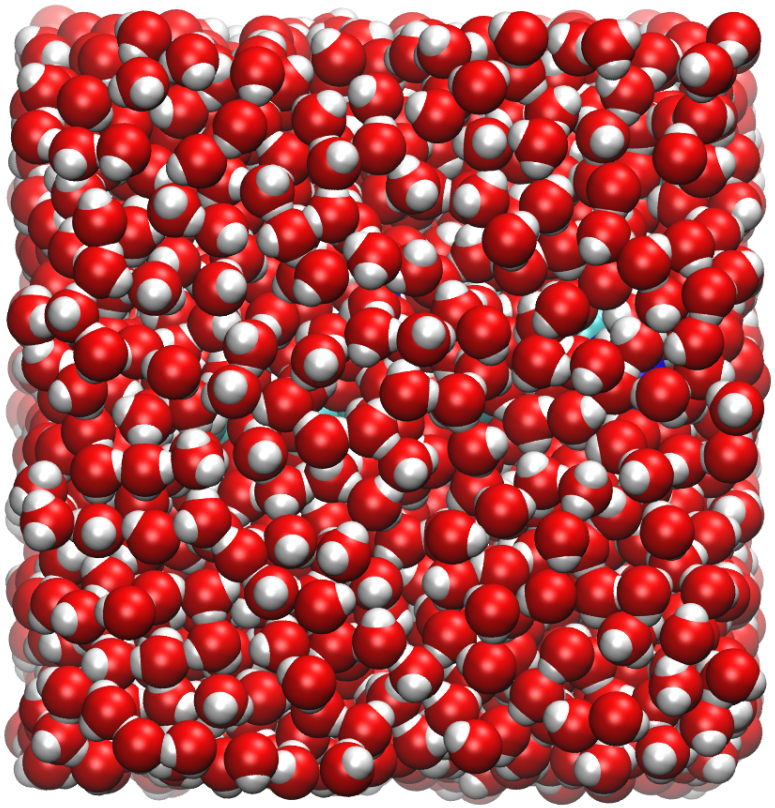




We want to simulate realistic 3D cellular microphysiology at length scales from *nm* and up and timescales of *ns* and longer.

Biomolecular Systems at the Molecular Level



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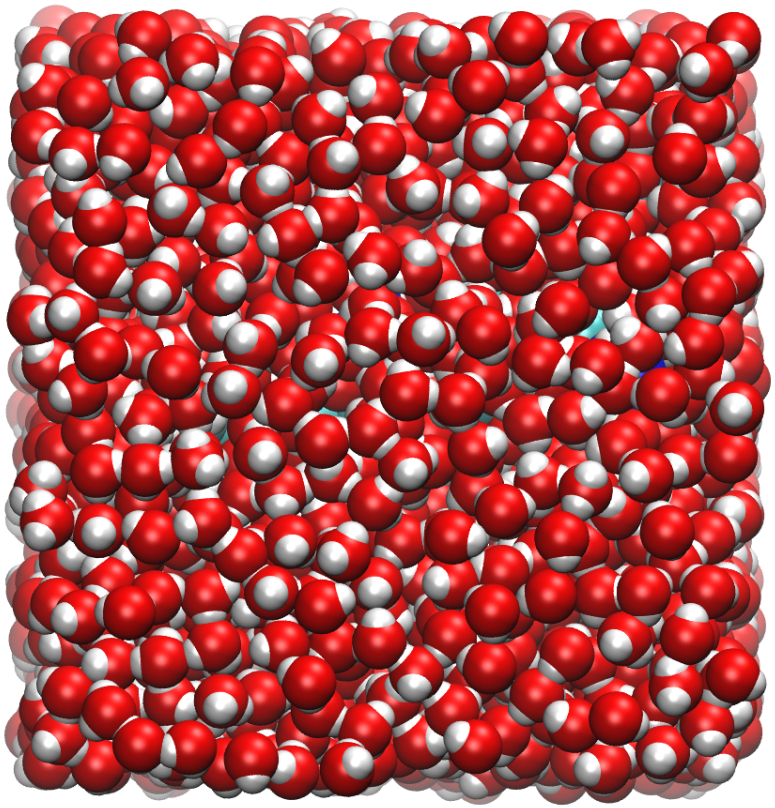

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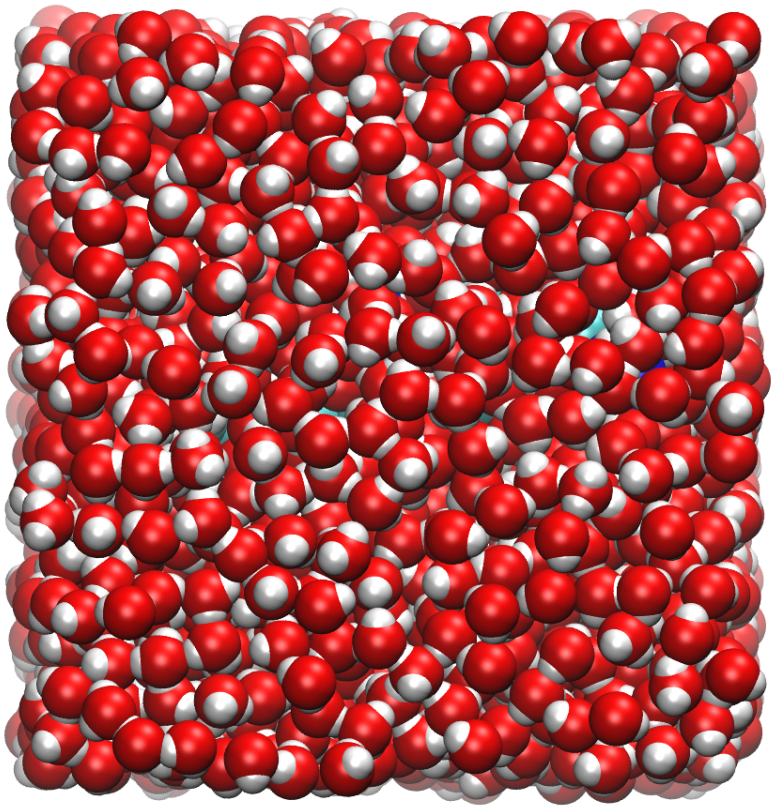

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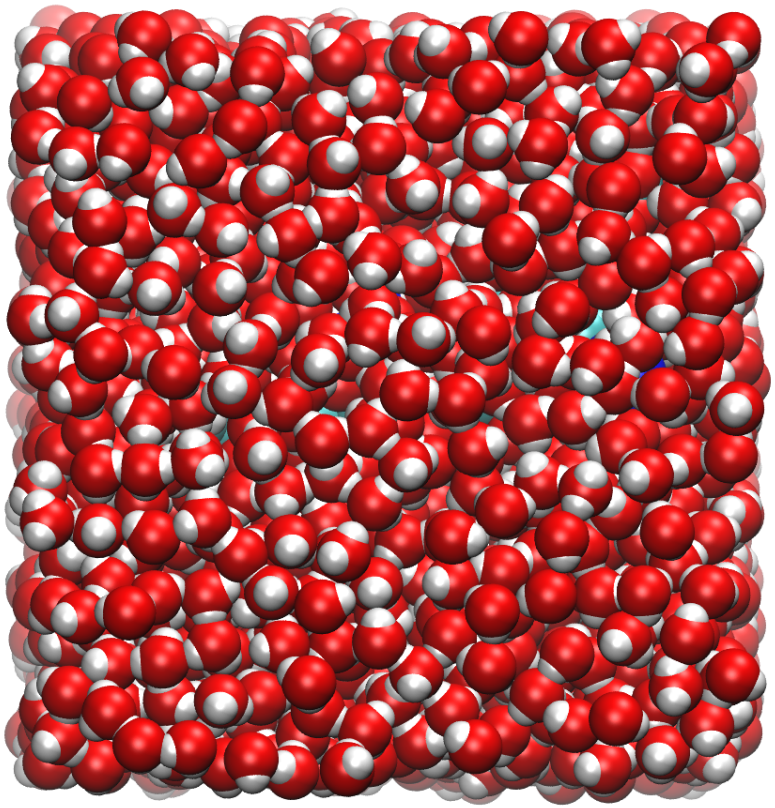

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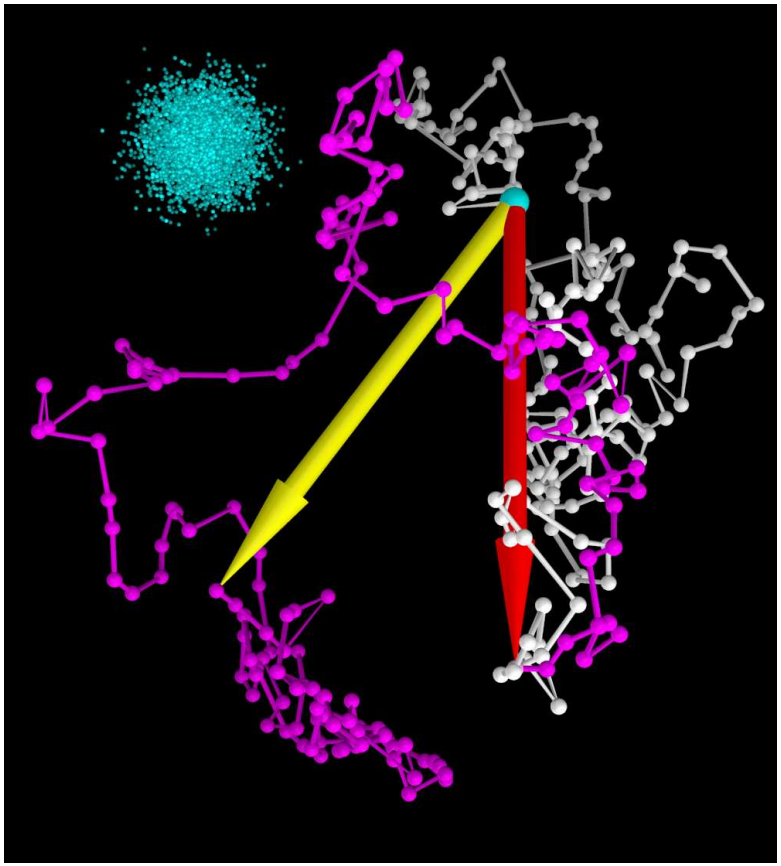
  $\bar{v} = \sqrt{\frac{3kT}{m}} = 640\ m/s$
- Frequent collisions of solute with water molecules randomizes their movement. Molecular motions are highly correlated.

Biomolecular Systems at the Microphysiological Level



- At the microphysiological level we are concerned with biology at time and length scales on the order of μs $(10^{-6}s)$ or longer and μm $(10^{-6}m)$ and up.
- At these scales molecular motion becomes uncorrelated and we can, to a first approximation, ignore water molecules and their rapid motion and instead describe molecular movement by stochastic *Brownian Motion* combined with a *diffusion coefficient*.
- Algorithms for simulations at the cellular level are typically based on PDE/finite element or stochastic methods.

Developed by physiologist Adolf Fick in 1855.

Fick's 1^{st} Law:

$$\mathbf{J}(\mathbf{r},t) = -D(C,t)\,\nabla C(\mathbf{r},t) \tag{1}$$

Fick's 2^{nd} Law:

$$\frac{\partial C(\mathbf{r},t)}{\partial t} = \nabla\left(D(C,t)\,\nabla C(\mathbf{r},t)\right) = D(C,t)\,\nabla^2 C(\mathbf{r},t) \tag{2}$$

Eq. 2 is called the **Diffusion Equation**.

Here, J, diffusion flux [Mol length^{-2} time^{-1}], D, diffusion coefficient [length^2 time^{-1}, $cm^2 s^{-1}$], C concentration [Mol length^{-3}, $mol\ l^{-1}$].

The solution to Fick's 2^{nd} Law provides the basis for MCell diffusion algorithm.

$$\frac{\partial C(\mathbf{r},t)}{\partial t} = D(C,t)\ \nabla^2 C(\mathbf{r},t) \tag{3}$$

In the neighborhood of a given molecule location, the concentration C can be assumed to be radially symmetric, $C(\mathbf{r},t) \equiv C(r,t)$ and Eq. 3 simplifies to

$$\frac{\partial C(r,t)}{\partial t} = D(C,t)\ \frac{\partial^2 C(r,t)}{\partial^2 r} \tag{4}$$

Equation can be solved analytically for certain boundary conditions. E.g. for a point source of M molecules the solution becomes

$$C(r,t) = \frac{M}{\lambda^3 \pi^{3/2}} e^{-r^2/\lambda^2} \quad , \quad \lambda = \sqrt{4Dt} \tag{5}$$

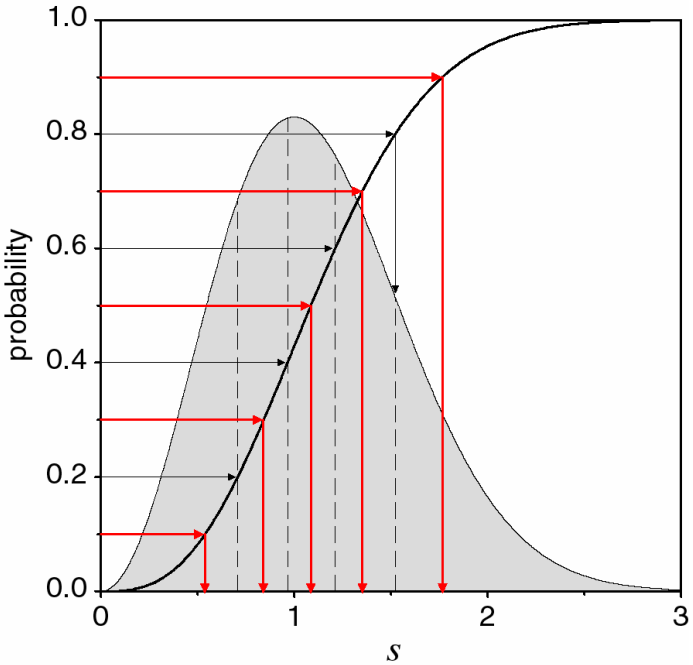
Eq. 5 can be directly converted into the fractional probability p_r for a displacement between r and $(r + dr)$ for a single diffusing molecule:

$$p_r = \frac{1}{\lambda^3 \pi^{3/2}} e^{-r^2/\lambda^2} (4\pi r^2) dr \tag{6}$$

$$p_s = \frac{4}{\sqrt{\pi}} s^2 e^{-s^2} ds \quad , \quad s = \frac{r}{\lambda} = \frac{r}{\sqrt{4Dt}} \tag{7}$$

Using Eq. 6 we can also compute the mean radial displacement $\bar{l}_r$

$$\bar{l}_r = \frac{2}{\pi}\lambda \sim \sqrt{t} \qquad (\bar{l}_\perp = \frac{\bar{l}_r}{2}) \tag{8}$$

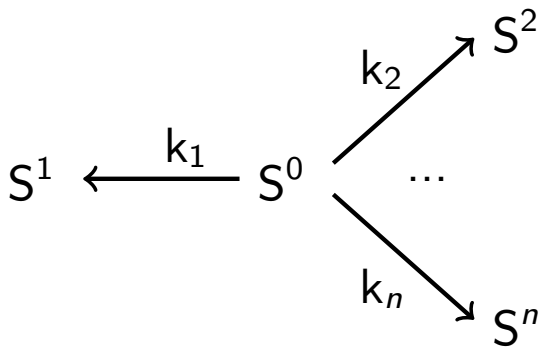


$$p_s = \frac{4}{\sqrt{\pi}} s^2 e^{-s^2} ds$$
$$s = \frac{r}{\sqrt{4Dt}}$$

To choose a radial distance R for diffusion we pick a random number X in $[0, 1]$ and solve

$$X = CDF(R) = \int_0^R p_s ds = erf(R) - \frac{2}{\sqrt{\pi}} R\ e^{-R^2} \qquad (9)$$

This can be efficiently computed during runtime of the simulation.



Unimolecular transition: Initial state S^0 can undergo one of n possible transitions to states S^1 through S^n with first order rate constants k_1, k_2, ... k_n.

We need to know the probability p_t that a single molecule in S^0 undergoes a transition. p_t is given by the fraction of $[S^0]$ that undergoes a transition during time t:

$$p_t = \frac{[S^1]_t + [S^2]_t + ... + [S^n]_t}{[S^0]_0} = 1 - \frac{[S^0]_t}{[S^0]_0} \tag{10}$$

From the rate equation we obtain

$$-d[S^0] = (k_1 + k_2 + ... + k_n)[S^0]dt = \left(\sum_{j=1}^{n} k_j\right)[S^0]dt \tag{11}$$

Eq. 11 can be integrated

$$\int_{[S^0]_0}^{[S^0]_t} \frac{d[S^0]}{[S^0]} = -\left(\sum_{j=1}^{n} k_j\right) \int_0^t dt \tag{12}$$

to yield

$$\frac{[S_0]_t}{[S_0]_0} = e^{-\sum_{j=1}^{n} k_j t} \tag{13}$$

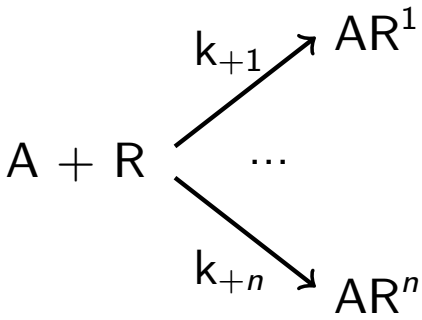
Substituting Eq. 13 into Eq. 10 then gives the probability p_t for unimolecular transitions as (here τ is the mean lifetime of S^0)

$$p_t = 1 - e^{-\sum_j k_j t} \quad , \qquad \tau = 1/\sum k_j \tag{14}$$

$$p_1 = p_t \frac{k_1}{\sum_j k_j}, \quad \ldots \quad p_n = p_t \frac{k_n}{\sum_j k_j}; \quad \sum_i p_i = p_t \tag{15}$$

Notes:

- The naïve way to choose unimolecular reactions is to compare a single random number in $[0, 1]$ to the cummulative probabilities $(p_{k1}, p_{k1} + p_{k2}, \ldots, 1)$.
- MCell3 instead computes the lifetime of each molecule from the exponential distribution of lifetimes $\rho(t) = \frac{1}{k} e^{-kt}$ and then uses its *scheduler* to schedule the unimolecular reaction to occur at the appropriate time.



Bimolecular Association: An example would be association between ligand A (volume molecule) and receptor R (surface molecule) with n possible binding sites.

We will derive a relation for p_b, the binding probability of ligand A to receptor R. The average rate of binding p_{bt} after N_H hits is given by

$$p_{bt} = 1 - (1 - p_b)^{N_H} \tag{16}$$

Next, we require that the *average binding rate* is equal to binding rate predicted by *mass action kinetics* given by

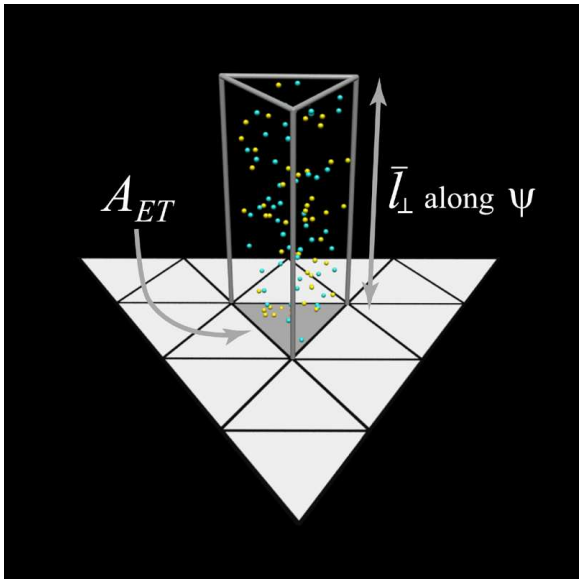
$$p_t = \sum_i k_{+i}[A]_0 \Delta t \quad , \quad \Delta t \to 0 \tag{17}$$

$$1 - (1 - p_b)^{N_h} = p_{bt} = p_t = \sum_i k_{+i}[A]_0 \Delta t \tag{18}$$

For small Δt, p_b and N_H approach zero and thus $(1 - p_b)^{N_H} \approx (1 - N_H p_b)$. Thus, Eq. 18 simplifies to

$$p_b = \sum_i k_{+i} \frac{[A]_0 \Delta t}{N_H} \quad , \quad \Delta t \to 0 \tag{19}$$

Next, we need to derive a relation for N_H the number of hits of A on R



The number of hits per unit time on a tile with surface area A_{ET} is given by

$$hits = 0.5 N_a \frac{\bar{l}_\perp}{\Delta t} A_{ET} [A]_0 \tag{20}$$

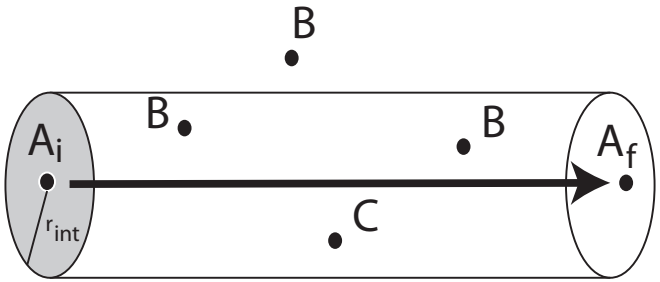
This results in $(\Delta t \to 0)$

$$N_H = \int_0^{\Delta t} hits \; dt \approx N_a A_{ET} [A]_0 \left(\frac{4D}{\pi}\right)^{1/2} \Delta t \tag{21}$$

Eliminating N_H in Eq. 19 with 21 then yields the final expression for p_b

$$p_b = \sum_i k_{+i} \frac{1}{2 N_a A_{ET}} \left(\frac{\pi \Delta t}{D}\right)^{1/2} \tag{22}$$

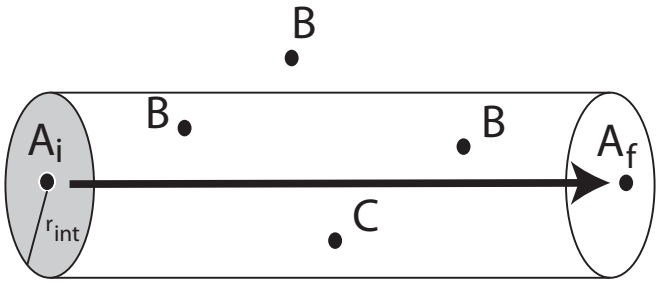
This can be efficiently computed at system initialization.



A + B --> *products*

- Volume molecules in MCell diffuse via ray-tracing along a randomly selected direction and diffusion step length computed as explained previously.

Monte Carlo Probabilities For Volume Reactions in MCell



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- Reaction partners are discovered and tested for reactions during ray marching. This unique approach provides good correlation between diffusive motion and location of reactants.

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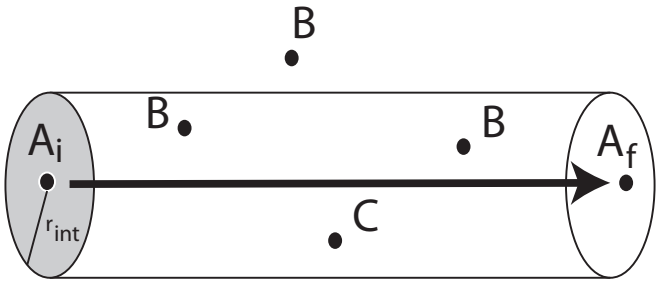

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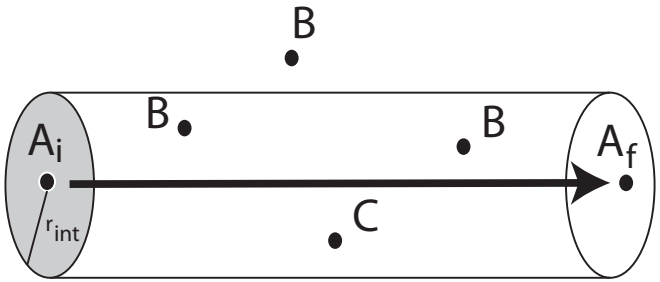

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Using an argument analogous to the one for bimolecular associations we can derive the following relation for the reaction probility between two diffusing volume molecules with diffusion constants $D1$ and $D2$:

$$p = \frac{k}{4A_{int}} \left(\frac{\pi \Delta t}{D_1 + D_2} \right)^{1/2} \tag{23}$$

References

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